

# LESSON PLAN

DEPARTMENT OF MATH AND SCIENCE , ITT, CHOUDWAR

SUBJECT: Engg.Chemistry

Periods: 4 per week

SEMESTER: 1<sup>st</sup> & 2<sup>nd</sup>

NAME OF FACULTY: Rajakrushna Nayak

ACADEMIC YEAR.2020-2021

Semester From date:

To Date:

No. of weeks: 15

Week	Class Day	Theory / Practical Topics
1 <sup>st</sup>	1 <sup>st</sup>	1. <b>Atomic structure</b> : Fundamental particles ( electron, proton & neutron Definition, mass and charge ).Rutherford's Atomic model ( postulates and failure), Atomic mass and mass number, Definition, examples and properties of Isotopes, isobars and isotones. Bohr's Atomic model ( Postulates only), Bohr-Bury scheme, Aufbau's principle, Hund's rule, Electronic configuration (up to atomic no 30).
	2 <sup>nd</sup>	2. <b>Chemical Bonding</b> : Definition , types ( Electrovalent, Covalent and Coordinate
2 <sup>nd</sup>	1 <sup>st</sup>	3. bond with examples ( formation of NaCl, MgCl <sub>2</sub> , H <sub>2</sub> ,Cl <sub>2</sub> , O <sub>2</sub> , N <sub>2</sub> , H <sub>2</sub> O, CH <sub>4</sub> , NH <sub>3</sub> , NH <sub>4</sub> <sup>+</sup> , SO <sub>2</sub> )
	2 <sup>nd</sup>	4. <b>Acid base theory</b> : Concept of Arrhenius, Lowry Bronsted and Lewis theory for acid and base with examples ( Postulates and limitations only)
3 <sup>rd</sup>	1 <sup>st</sup>	5. Neutralization of acid & base. Definition of Salt, Types of salts ( Normal, acidic, basic, double, complex and mixed salts, definitions with 2 examples from each).
	2 <sup>nd</sup>	6. <b>Solutions</b> : Definitions of atomic weight, molecular weight, Equivalent weight. Determination of equivalent weight of Acid, Base and Salt.
4 <sup>th</sup>	1 <sup>st</sup>	7. Modes of expression of the concentrations ( Molarity , Normality & Molality) with Simple Problems. pH of solution ( definition with simple numericals ) Importance of pH in industry ( sugar, textile, paper industries only)
	2 <sup>nd</sup>	8. <b>Electrochemistry</b> : Definition and types ( Strong & weak) of Electrolytes with example. Electrolysis ( Principle & process) with example of NaCl (fused and aqueous solution).
5 <sup>th</sup>	1 <sup>st</sup>	9. Faraday's 1 <sup>st</sup> and 2 <sup>nd</sup> law of Electrolysis ( Statement, mathematical expression and Simple numerical) Industrial application of Electrolysis- Electroplating ( Zinc only).

	2 <sup>nd</sup>	10. <b>Corrosion:</b> Definition of Corrosion, Types of Corrosion- Atmospheric Corrosion
6 <sup>th</sup>	1 <sup>st</sup>	11. Waterline corrosion. Mechanism of rusting of Iron only. Protection from Corrosion by (i) Alloying and (ii) Galvanization.
	2 <sup>nd</sup>	12. <b>Metallurgy:</b> Definition of Mineral, ores , gangue with example. Distinction between Ores And Minerals
7 <sup>th</sup>	1 <sup>st</sup>	13. General methods of extraction of metals,
	2 <sup>nd</sup>	14. i) Ore Dressing ii) Concentration ( Gravity separation, magnetic separation, Froth floatation & leaching) iii) Oxidation (Calcinations, Roasting ) iv) Reduction (Smelting, Definition & examples of flux, slag) v) Refining of the metal ( Electro refining, & Distillation only)
8 <sup>th</sup>	1 <sup>st</sup>	15. <b>Alloys:</b> Definition of alloy. Types of alloys ( Ferro, Non Ferro & Amalgam) with example. Composition and uses of Brass, Bronze, Alnico, Duralumin
	2 <sup>nd</sup>	16. <b>Hydrocarbons :</b> Saturated and Unsaturated Hydrocarbons ( Definition with example)
9 <sup>th</sup>	1 <sup>st</sup>	17. Aliphatic and Aromatic Hydrocarbons ( Huckle's rule only). Difference between Aliphatic and aromatic hydrocarbons
	2 <sup>nd</sup>	18. IUPAC system of nomenclature of Alkane, Alkene, Alkyne, alkyl halide and alcohol ( up to 6 carbons ) with bond line notation.
10 <sup>th</sup>	1 <sup>st</sup>	19. Uses of some common aromatic compounds ( Benzene, Toluene, BHC, Phenol, Naphthalene, Anthracene and Benzoic acid) in daily life.
	2 <sup>nd</sup>	20. <b>Water Treatment :</b> Sources of water, Soft water, Hard water, hardness, types of Hardness (temporary or carbonate and permanent or non-carbonate),
11 <sup>th</sup>	1 <sup>st</sup>	21. Removal of hardness by lime soda method ( hot lime & cold lime—Principle, process & advantages)
	2 <sup>nd</sup>	22. Advantages of Hot lime over cold lime process. Organic Ion exchange method ( principle, process, and regeneration of exhausted resins)
12 <sup>th</sup>	1 <sup>st</sup>	23. . <b>Lubricants:</b> Definition of lubricant, Types ( solid, liquid and semisolid with
	2 <sup>nd</sup>	24. examples only ) and specific uses of lubricants ( Graphite, Oils, Grease), Purpose of lubrication

13 <sup>th</sup>	1 <sup>st</sup>	25. <b>Fuel:</b> Definition and classification of fuel, Definition of calorific value of fuel, Choice of good fuel.
	2 <sup>nd</sup>	26. Liquid: Diesel, Petrol, and Kerosene --- Composition and uses. Gaseous: Producer gas and Water gas (Composition and uses). Elementary idea about LPG, CNG and coal gas (Composition and uses only).
14 <sup>th</sup>	1 <sup>st</sup>	27. <b>Polymer:</b> Definition of Monomer, Polymer, Homo-polymer, Co-polymer and Degree of polymerization. Difference between Thermosetting and Thermoplastic, Composition and uses of Polythene, & Poly-Vinyl Chloride and Bakelite.
	2 <sup>nd</sup>	28. Definition of Elastomer ( Rubber). Natural Rubber (it's draw backs ). Vulcanisation of Rubber. Advantages of Vulcanised rubber over raw rubber.
15 <sup>th</sup>	1 <sup>st</sup>	29. <b>Chemicals in Agriculture:</b> Pesticides: Insecticides, herbicides, fungicides- Examples and uses.
	2 <sup>nd</sup>	30. Bio Fertilizers: Definition, examples and uses.

# LESSON PLAN

DEPARTMENT OF MATH AND SCIENCE , ITT, CHOUDWAR

SUBJECT: ENGG. Engineering Chemistry Lab    Periods: 4 per week    SEMESTER: 1<sup>st</sup> & 2<sup>nd</sup>

NAME OF FACULTY: Rajakrushna Nayak    ACADEMIC YEAR.2020-2021

Semester From date:                      To Date:                                      No. of weeks: 15

Week	Class Day	Theory / Practical Topics
1st	1 <sup>st</sup>	1. Preparation and study of physical and chemical properties CO <sub>2</sub> gas.
	2 <sup>nd</sup>	2. Preparation and study of physical and chemical properties CO <sub>2</sub> gas.
2 <sup>nd</sup>	1 <sup>st</sup>	3. Preparation and study of physical and chemical properties CO <sub>2</sub> gas.
	2 <sup>nd</sup>	4. Preparation and study of physical and chemical properties NH <sub>3</sub> gas.
3 <sup>rd</sup>	1 <sup>st</sup>	5. Preparation and study of physical and chemical properties NH <sub>3</sub> gas.
	2 <sup>nd</sup>	6. Preparation and study of physical and chemical properties NH <sub>3</sub> gas.
4 <sup>th</sup>	1 <sup>st</sup>	7. Crystallization of Copper sulphate from copper carbonate
	2 <sup>nd</sup>	8. Crystallization of Copper sulphate from copper carbonate
5 <sup>th</sup>	1 <sup>st</sup>	9. Crystallization of Copper sulphate from copper carbonate
	2 <sup>nd</sup>	10. Simple acid-base titrations (i) Acidimetry (ii) Alkalimetry
6 <sup>th</sup>	1 <sup>st</sup>	11. Simple acid-base titrations (i) Acidimetry (ii) Alkalimetry
	2 <sup>nd</sup>	12. Simple acid-base titrations (i) Acidimetry (ii) Alkalimetry
7 <sup>th</sup>	1 <sup>st</sup>	13. Tests for acid radicals (Known): (i) Carbonate,
	2 <sup>nd</sup>	14. Tests for acid radicals (Known): (ii) Sulphide,

8 <sup>th</sup>	1 <sup>st</sup>	15. Tests for acid radicals (Known): (iii) Chloride,
	2 <sup>nd</sup>	16. Tests for acid radicals (Known): (iv) Nitrate
9 <sup>th</sup>	1 <sup>st</sup>	17. Tests for acid radicals (Known): (v) Sulphate.
	2 <sup>nd</sup>	18. Tests for acid radicals (Known): (i) Carbonate,
10 <sup>th</sup>	1 <sup>st</sup>	19. Tests for acid radicals (Known): (ii) Sulphide,
	2 <sup>nd</sup>	20. Tests for acid radicals (Known): (iii) Chloride,
11 <sup>th</sup>	1 <sup>st</sup>	21. Tests for acid radicals (Known): (iv) Nitrate
	2 <sup>nd</sup>	22. Tests for acid radicals (Known): (v) Sulphate.
12 <sup>th</sup>	1 <sup>st</sup>	23. Test for unknown Acid radicals
	2 <sup>nd</sup>	24. Test for unknown Acid radicals
13 <sup>th</sup>	1 <sup>st</sup>	25. Test for unknown Acid radicals
	2 <sup>nd</sup>	26. Test for unknown basic radicals
14 <sup>th</sup>	1 <sup>st</sup>	27. Test for unknown basic radicals
	2 <sup>nd</sup>	28. Test for unknown basic radicals
15 <sup>th</sup>	1 <sup>st</sup>	29. Test for unknown salt (composed of one basic radical and one acid radical)
	2 <sup>nd</sup>	30. Test for unknown salt (composed of one basic radical and one acid radical)

# LESSON PLAN

DEPARTMENT OF MATH & SCIENCE ENGINEERING, ITT, CHOUDWAR

SUBJECT: Communicative English

Periods: 4 per week

SEMESTER: 1<sup>st</sup>/2<sup>nd</sup>

NAME OF FACULTY: SANGITA MUDULI

ACADEMIC YEAR.2020-2021

Semester From date:

To Date:

No. of weeks: 15

Week	Class Day	Practical Topics
1st	1 <sup>st</sup>	1. Listening Skills(1)
	2 <sup>nd</sup>	1. Listening Skills(2)
2 <sup>nd</sup>	1 <sup>st</sup>	2. Speaking Skill
	2 <sup>nd</sup>	2. Speaking Skill
3 <sup>rd</sup>	1 <sup>st</sup>	3. Personality Development
	2 <sup>nd</sup>	3. Personality Development
4 <sup>th</sup>	1 <sup>st</sup>	4. Interpersonal Skills
	2 <sup>nd</sup>	4. Interpersonal Skills
5 <sup>th</sup>	1 <sup>st</sup>	5. Presenting in GD, Seminars and Conferences.
	2 <sup>nd</sup>	5. Presenting in GD, Seminars and Conferences.
		REVISION
6 <sup>th</sup>	1 <sup>st</sup>	1. Listening Skills
	2 <sup>nd</sup>	1. Listening Skills
7 <sup>th</sup>	1 <sup>st</sup>	2. Speaking Skill
	2 <sup>nd</sup>	2. Speaking Skill

8 <sup>th</sup>	1 <sup>st</sup>	3. Personality Development
	2 <sup>nd</sup>	3. Personality Development
9 <sup>th</sup>	1 <sup>st</sup>	4. Interpersonal Skills
	2 <sup>nd</sup>	4. Interpersonal Skills
10 <sup>th</sup>	1 <sup>st</sup>	5. Presenting in GD, Seminars and Conferences.
	2 <sup>nd</sup>	5. Presenting in GD, Seminars and Conferences.
11 <sup>th</sup>	1 <sup>st</sup>	5. Presenting in GD, Seminars and Conferences.
	2 <sup>nd</sup>	4. Interpersonal Skills
12 <sup>th</sup>	1 <sup>st</sup>	3. Personality Development Initiation Physical Appearance Audience Purpose
	2 <sup>nd</sup>	2. Speaking Skill Reading aloud of dialogues, texts, poems, speeches focusing on intonation. <ul style="list-style-type: none"> <li>• Self-introduction</li> <li>• Role plays on any two-situations.</li> <li>• Telephonic Conversations</li> </ul>
13 <sup>th</sup>	1 <sup>st</sup>	1. Listening Skills The student should be able to listen to s text read aloud in normal speed with focus on intonation. After listening the student can fill-in-blanks, choose a suitable title, make a summary, supply required information and be able to answer comprehension questions from the passage read aloud.
	2 <sup>nd</sup>	1. Listening Skills The student should be able to listen to s text read aloud in normal speed with focus on intonation. After listening the student can fill-in-blanks, choose a suitable title, make a summary, supply required information and be able to answer comprehension questions from the passage read aloud.
14 <sup>th</sup>	1 <sup>st</sup>	2. Speaking Skill Reading aloud of dialogues, texts, poems, speeches focusing on intonation. <ul style="list-style-type: none"> <li>• Self-introduction</li> <li>• Role plays on any two-situations.</li> <li>• Telephonic Conversations</li> </ul>

	2 <sup>nd</sup>	3. Personality Development Initiation Physical Appearance  Audience Purpose
15 <sup>th</sup>	1 <sup>st</sup>	4. Interpersonal Skills Appropriate use of non-verbal skills in face to face communication [i.e. Viva-Voce, group -interviews, GDs and seminars.]
	2 <sup>nd</sup>	5. Presenting in GD, Seminars and Conferences. • Leadership Quality • Time Management • Achieving the target,



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DEPARTMENT OF MATH & SCIENCE ENGINEERING, ITT, CHOUDWAR

SUBJECT: Communicative English

Periods: 4 per week

SEMESTER: 1<sup>st</sup>/2<sup>nd</sup>

NAME OF FACULTY: SANGITA MUDULI

ACADEMIC YEAR.2020-2021

Semester From date:

To Date:

No. of weeks: 15

Week	Class Day	Theory Topics
1st	1 <sup>st</sup>	1.Unit-1 literature appreciation A reading comprehension
	2 <sup>nd</sup>	2. Unit-1 literature appreciation A reading comprehension
2 <sup>nd</sup>	1 <sup>st</sup>	B.Text standing up your self BY YEVGENY YEVTUSHENKO
	2 <sup>nd</sup>	B.Text standing up your self BY YEVGENY YEVTUSHENKO
3 <sup>rd</sup>	1 <sup>st</sup>	3. The Magic of teamwork BY SAM PITRODA
	2 <sup>nd</sup>	The Magic of teamwork BY SAM PITRODA
4 <sup>th</sup>	1 <sup>st</sup>	4. Inchcape Rock BY Robert southey
	2 <sup>nd</sup>	Inchcape Rock BY Robert southey
5 <sup>th</sup>	1 <sup>st</sup>	TO my true friend BY Elizabeth pinard
	2 <sup>nd</sup>	TO my true friend BY Elizabeth pinard
6 <sup>th</sup>	1 <sup>st</sup>	Unit 2 use of synonyms ,antonyms
	2 <sup>nd</sup>	Articles and Determiners
7 <sup>th</sup>	1 <sup>st</sup>	Modal verbs
	2 <sup>nd</sup>	Tenses,
8 <sup>th</sup>	1 <sup>st</sup>	Voice change

	2 <sup>nd</sup>	Subject verb agreement
9 <sup>th</sup>	1 <sup>st</sup>	A paragraph writing , meaning
	2 <sup>nd</sup>	Features of paragraph writing
10 <sup>th</sup>	1 <sup>st</sup>	Notice
	2 <sup>nd</sup>	agenda
11 <sup>th</sup>	1 <sup>st</sup>	D.Report writing
	2 <sup>nd</sup>	Writing personal letter
12 <sup>th</sup>	1 <sup>st</sup>	Writing business letters
	2 <sup>nd</sup>	<i>Unit v – elements of communication , meaning of communication</i>
13 <sup>th</sup>	1 <sup>st</sup>	Communication of model , process of communication
	2 <sup>nd</sup>	Professional communication , types of professional communication
14 <sup>th</sup>	1 <sup>st</sup>	Formal communication , informal communication
	2 <sup>nd</sup>	Meaning of non verbal communication
15 <sup>th</sup>	1 <sup>st</sup>	Different areas of non verbal communication
	2 <sup>nd</sup>	Language of signs and symbols

# LESSON PLAN

DEPARTMENT OF MATH & SCIENCE, ITT, CHOUDWAR

SUBJECT: ENGG. MATH II

Periods: 6 per week

SEMESTER: 2<sup>nd</sup>

NAME OF FACULTY: Sk. S. Ali

ACADEMIC YEAR.2020-2021

Semester From date:

To Date:

No. of weeks: 15

Week	Class Day	Theory / Practical Topics
1st	1 <sup>st</sup>	<b>1) VECTOR ALGEBRA</b> a) Introduction
	2 <sup>nd</sup>	b) Types of vectors (null vector, parallel vector , collinear vectors) (in component form )
	3 <sup>rd</sup>	c) Representation of vector
2 <sup>nd</sup>	1 <sup>st</sup>	d) Magnitude and direction of vectors
	2 <sup>nd</sup>	e) Addition and subtraction of vectors
	3 <sup>rd</sup>	f) Position vector
3 <sup>rd</sup>	1 <sup>st</sup>	g) Scalar product of two vectors
	2 <sup>nd</sup>	h) Geometrical meaning of dot product
	3 <sup>rd</sup>	i) Angle between two vectors
4 <sup>th</sup>	1 <sup>st</sup>	j) Scalar and vector projection of two vectors
	2 <sup>nd</sup>	k) Vector product and geometrical meaning (Area of triangle and parallelogram)
	3 <sup>rd</sup>	<b>2) LIMITS AND CONTINUITY</b> a) Definition of function, based on set theory
5 <sup>th</sup>	1 <sup>st</sup>	b) Types of functions i) Constant function ii) Identity function iii) Absolute value function
	2 <sup>nd</sup>	iv)The Greatest integer function

	3 <sup>rd</sup>	v) Trigonometric function
6 <sup>th</sup>	1 <sup>st</sup>	vi) Exponential function
	2 <sup>nd</sup>	vii) Logarithmic function
	3 <sup>rd</sup>	c) Introduction of limit
7 <sup>th</sup>	1 <sup>st</sup>	d) Existence of limit
	2 <sup>nd</sup>	e) Methods of evaluation of limit  i) $\lim_{x \rightarrow 0} \frac{x^n - a^n}{x - a} = na^{n-1}$ ii) $\lim_{x \rightarrow 0} \frac{a^x - 1}{x} = \log_e a$ iii) $\lim_{x \rightarrow 0} \frac{e^x - 1}{x} = 1$ iv) $\lim_{x \rightarrow 0} (1 + x)^{1/x} = e$ v) $\lim_{x \rightarrow \infty} \left(1 + \frac{1}{x}\right)^x = e$ vi) $\lim_{x \rightarrow 0} \frac{\log(1+x)}{x} = 1$ vii) $\lim_{x \rightarrow 0} \frac{\sin x}{x} = 1$ viii) $\lim_{x \rightarrow 0} \frac{\tan x}{x} = 1$
	3 <sup>rd</sup>	f) Definition of continuity of a function at a point and problems based on it
8 <sup>th</sup>	1 <sup>st</sup>	a) Derivative of a function at a point
	2 <sup>nd</sup>	b) Algebra of derivative
	3 <sup>rd</sup>	c) Derivative of standard functions
9 <sup>th</sup>	1 <sup>st</sup>	d) Derivative of composite function (Chain Rule )
	2 <sup>nd</sup>	e) Methods of differentiation of i) Parametric function
	3 <sup>rd</sup>	ii) Implicit function
10 <sup>th</sup>	1 <sup>st</sup>	iii) Logarithmic function

	2 <sup>nd</sup>	iv) a function with respect to another function
	3 <sup>rd</sup>	f) Applications of Derivative
11 <sup>th</sup>	1 <sup>st</sup>	i) Successive Differentiation (up to second order)
	2 <sup>nd</sup>	ii) Partial Differentiation (function of two variables up to second order)
	3 <sup>rd</sup>	g) Problems based on above
12 <sup>th</sup>	1 <sup>st</sup>	<b>4) INTEGRATION</b> a) Definition of integration as inverse of differentiation
	2 <sup>nd</sup>	Definition of integration as inverse of differentiation
	3 <sup>rd</sup>	b) Integrals of standard functions
13 <sup>th</sup>	1 <sup>st</sup>	c) Methods of integration
	2 <sup>nd</sup>	
		i) Integration by substitution
	ii) Integration by parts	
14 <sup>th</sup>	1 <sup>st</sup>	d) Integration of the following forms i) $\int \frac{dx}{x^2+a^2}$ ii) $\int \frac{dx}{x^2-a^2}$ iii) $\int \frac{dx}{a^2-x^2}$ iv) $\int \frac{dx}{\sqrt{x^2+a^2}}$ v) $\int \frac{dx}{\sqrt{x^2-a^2}}$ vi) $\int \frac{dx}{\sqrt{a^2-x^2}}$ vii) $\int \frac{dx}{x\sqrt{x^2-a^2}}$ viii) $\int \sqrt{a^2-x^2} dx$ ix) $\int \sqrt{a^2+x^2} dx$ x) $\int \sqrt{x^2-a^2} dx$
		Definite integral, properties of definite integrals i) $\int_0^a f(x) dx = \int_0^a f(a-x) dx$ ii) $\int_a^b f(x) dx = -\int_b^a f(x) dx$ iii) $\int_a^c f(x) dx = \int_a^b f(x) dx + \int_b^c f(x) dx$ , $a < b < c$ iv) $\int_{-a}^a f(x) dx = 0$ , if $f(x) = \text{odd}$ $= 2\int_0^a f(x) dx$ , if $f(x) = \text{even}$

	2 <sup>nd</sup>	ii) Area of a circle with centre at origin
	3 <sup>rd</sup>	f) Application of integration
	1 <sup>st</sup>	i) Area enclosed by a curve and X – axis
15 <sup>th</sup>	2 <sup>nd</sup>	<b>5) DIFFERENTIAL EQUATION</b> a) Order and degree of a differential equation
	3 <sup>rd</sup>	b) Solution of differential equation i) 1st order and 1st degree equation by the method of separation of variables

# LESSON PLAN

DEPARTMENT OF MATH & SCIENCE, ITT, CHOUDWAR

SUBJECT: ENGG. MATH I

Periods: 6 per week

SEMESTER: 1<sup>st</sup>

NAME OF FACULTY: Sk. S. Ali

ACADEMIC YEAR.2020-2021

Semester From date:

To Date:

No. of weeks: 15

Week	Class Day	Theory / Practical Topics
1 <sup>st</sup>	1 <sup>st</sup>	<b>INTRODUCTION CLASS</b>
	2 <sup>nd</sup>	<b>1) MATRICES AND DETERMINANTS</b> 2) Types of matrices
	3 <sup>rd</sup>	a)Types of matrices
2 <sup>nd</sup>	1 <sup>st</sup>	b)Algebra of matrices
	2 <sup>nd</sup>	3) Algebra of matrices
	3 <sup>rd</sup>	4) Determinant
3 <sup>rd</sup>	1 <sup>st</sup>	c)Determinant
	2 <sup>nd</sup>	d) Properties of determinant
	3 <sup>rd</sup>	5) Properties of determinant
4 <sup>th</sup>	1 <sup>st</sup>	6) Inverse of a matrix (second and third order) (Question should be on second order matrix)
	2 <sup>nd</sup>	e) Inverse of a matrix (second and third order) (Question should be on second order matrix)
	3 <sup>rd</sup>	f) Cramer's Rule (Question should be on two variables)
5 <sup>th</sup>	1 <sup>st</sup>	7) Cramer's Rule (Question should be on two variables)
	2 <sup>nd</sup>	g) Cramer's Rule (Question should be on two variables)

	3 <sup>rd</sup>	h) Solution of simultaneous equations by matrix inverse method (Question should be on two variables)
6 <sup>th</sup>	1 <sup>st</sup>	i) Solution of simultaneous equations by matrix inverse method (Question should be on two variables)
	2 <sup>nd</sup>	<b>2) TRIGONOMETRY</b> a) Trigonometrically ratios
	3 <sup>rd</sup>	b) Compound angles, multiple and sub-multiple angles (only formulae)
7 <sup>th</sup>	1 <sup>st</sup>	c) Define inverse circular functions and its properties (no derivation)
	2 <sup>nd</sup>	<b>3) CO-ORDINATE GEOMETRY IN TWO DIMENSIONS (Straight line)</b> a) Introduction of geometry in two dimension
	3 <sup>rd</sup>	b) Distance formulae, division formulae, area of a triangle (only formulae no derivation)
8 <sup>th</sup>	1 <sup>st</sup>	c) Define slope of a line, angle between two lines (only F), condition of perpendicularity and parallelism.
	2 <sup>nd</sup>	d) Different forms of straight lines (only formulae)
	3 <sup>rd</sup>	i) One point form
9 <sup>th</sup>	1 <sup>st</sup>	ii) two point form
	2 <sup>nd</sup>	iii) slope form
	3 <sup>rd</sup>	iv) intercept form
10 <sup>th</sup>	1 <sup>st</sup>	v) Perpendicular form
	2 <sup>nd</sup>	e) Equation of a line passing through a point and



	3 <sup>rd</sup>	i) parallel to a line
11 <sup>th</sup>	1 <sup>st</sup>	ii) Perpendicular to a line
	2 <sup>nd</sup>	f) Equation of a line passing through the intersection of two lines
	3 <sup>rd</sup>	g) Distance of a point from a line
12 <sup>th</sup>	1 <sup>st</sup>	<b>4) CIRCLE</b> a) Equation of a circle i) center radius form
	2 <sup>nd</sup>	ii) general equation of a circle
	3 <sup>rd</sup>	iii) end point of diameter form
13 <sup>th</sup>	1 <sup>st</sup>	<b>5) CO-ORDINATE GEOMETRY IN THREE DIMENSIONS</b> a) Distance formulae, section formulae, direction ratio
	2 <sup>nd</sup>	b) direction cosine, angle between two lines
		c) (condition of parallelism and perpendicularity)
14 <sup>th</sup>	1 <sup>st</sup>	d) Equation of a plane General form, angle between two planes, perpendicular distance of a point from a plane, equation of a plane passing through a point and
	2 <sup>nd</sup>	e) parallel to a plane
	3 <sup>rd</sup>	f) perpendicular to a plane
15 <sup>th</sup>	1 <sup>st</sup>	<b>6) SPHERE</b> a) Equation of a sphere i) center radius form
	2 <sup>nd</sup>	ii) general form
	3 <sup>rd</sup>	iii) two end points of a diameter form (only formulae and problems)

# LESSON PLAN

DEPARTMENT OF MATH AND SCIENCE , ITT, CHOUDWAR

SUBJECT: ENGG. PHYSICS THEORY

Periods: 4 per week

SEMESTER: 1<sup>st</sup> & 2<sup>nd</sup>

NAME OF FACULTY: Pranabesh Bhanja

ACADEMIC YEAR.2020-2021

Semester From date:

To Date:

No. of weeks: 15

Week	Class Day	Theory / Practical Topics
1 <sup>st</sup>	1 <sup>st</sup>	1. Unit And Dimensions, Physical quantities - (Definition), Definition of fundamental and derived units, systems of units (FPS, CGS, MKS and SI units)
	2 <sup>nd</sup>	2. Definition of dimension and Dimensional formulae of physical quantities, Dimensional equations
2 <sup>nd</sup>	1 <sup>st</sup>	3. Principle of homogeneity, Checking the dimensional correctness of Physical relations.
	2 <sup>nd</sup>	4. Scalar and Vector quantities (definition and concept), Representation of a Vector – examples, types of vectors.
3 <sup>rd</sup>	1 <sup>st</sup>	5. Triangle and Parallelogram law of vector Addition (Statement only), Simple Numerical
	2 <sup>nd</sup>	6. Resolution of Vectors – Simple Numericals on Horizontal and Vertical components, Vector multiplication (scalar product and vector product of vectors)
4 <sup>th</sup>	1 <sup>st</sup>	7. Concept of Rest and Motion, Displacement, Speed, Velocity, Acceleration & FORCE (Definition, formula, dimension & SI units)
	2 <sup>nd</sup>	8. Equations of Motion under Gravity (upward and downward motion) - no derivation
5 <sup>th</sup>	1 <sup>st</sup>	9. Circular motion: Angular displacement, Angular velocity and Angular acceleration (definition, formula & SI units), Relation between –(i) Linear & Angular velocity, (ii) Linear & Angular acceleration)
	2 <sup>nd</sup>	10. Define Projectile, Examples of Projectile
6 <sup>th</sup>	1 <sup>st</sup>	11. Expression for Equation of Trajectory, Time of Flight, Maximum Height and Horizontal Range for a projectile fired at an angle, Condition for maximum Horizontal Range
	2 <sup>nd</sup>	12. Work – Definition, Formula & SI units, Friction – Definition & Concept, Types of friction (static, dynamic)
7 <sup>th</sup>	1 <sup>st</sup>	13. Limiting Friction (Definition with Concept), Laws of Limiting Friction (Only statement, No Experimental Verification)
	2 <sup>nd</sup>	14. Coefficient of Friction – Definition & Formula, Simple Numericals, Methods to reduce friction

8 <sup>th</sup>	1 <sup>st</sup>	15. Newton's Laws of Gravitation – Statement and Explanation, Universal Gravitational Constant (G)- Definition, Unit and Dimension
	2 <sup>nd</sup>	16. Acceleration due to gravity (g)- Definition and Concept, Definition of mass and weight, Relation between g and G
9 <sup>th</sup>	1 <sup>st</sup>	17. Variation of g with altitude and depth (No derivation – Only Explanation), Kepler's Laws of Planetary Motion (Statement only).
	2 <sup>nd</sup>	18. Simple Harmonic Motion (SHM) - Definition & Examples, Expression (Formula/Equation) for displacement
10 <sup>th</sup>	1 <sup>st</sup>	19. Velocity, acceleration of a body/ particle in SHM, Wave motion – Definition & Concept, Transverse and Longitudinal wave motion – Definition, Examples & Comparison
	2 <sup>nd</sup>	20. Definition of different wave parameters (Amplitude, Wavelength, Frequency, Time Period, Derivation of Relation between Velocity, Frequency and Wavelength of a wave, Ultrasonics – Definition, Properties & Applications.
11 <sup>th</sup>	1 <sup>st</sup>	21. Heat and Temperature – Definition & Difference, Units of Heat (FPS, CGS, MKS & SI), Specific Heat (concept, definition, unit, dimension and simple numerical), Change of state (concept), Latent Heat (concept, definition, unit, dimension and simple numerical)
	2 <sup>nd</sup>	22. Thermal Expansion – Definition & Concept, Expansion of Solids (Concept), Coefficient of linear, superficial and cubical expansions of Solids – Definition & Units, Relation between $\alpha$ , $\beta$ & $\gamma$ , Work and Heat - Concept & Relation, Joule's Mechanical Equivalent of Heat (Definition, Unit), First Law of Thermodynamics (Statement and concept only)
12 <sup>th</sup>	1 <sup>st</sup>	23. Reflection & Refraction – Definition, Laws of reflection and refraction (Statement only), Refractive index – Definition, Formula & Simple numerical, Critical Angle and Total internal reflection – Concept, Definition & Explanation, Refraction through Prism (Ray Diagram & Formula only – NO derivation), Fiber Optics – Definition, Properties & Applications
	2 <sup>nd</sup>	24. Electrostatics – Definition & Concept, Statement & Explanation of Coulombs laws, Definition of Unit charge, Absolute & Relative Permittivity ( $\epsilon$ ) – Definition, Relation & Unit. Electrostatics – Definition & Concept, Statement & Explanation of Coulombs laws, Definition of Unit charge, Absolute & Relative Permittivity ( $\epsilon$ ) – Definition, Relation & Unit
13 <sup>th</sup>	1 <sup>st</sup>	25. Electric potential and Electric Potential difference (Definition, Formula & SI Units), Electric field, Electric field intensity (E) – Definition, Formula & Unit, Capacitance - Definition, Formula & Unit, Series and Parallel combination of Capacitors (No derivation, Formula for effective/Combined/total capacitance & Simple numericals).
	2 <sup>nd</sup>	26. Magnet, Properties of a magnet, Coulomb's Laws in Magnetism – Statement & Explanation, Unit Pole (Definition), Magnetic field, Magnetic Field intensity

		(H), (Definition, Formula & SI Unit),Magnetic lines of force ( Definition and Properties),Magnetic Flux ( $\Phi$ ) & Magnetic Flux Density (B) – Definition, Formula & Unit
14 <sup>th</sup>	1 <sup>st</sup>	27.Electric Current – Definition, Formula & SI Units,Ohm’s law and its applications,Series and Parallel combination of resistors (No derivation, Formula for effective/Combined/ total resistance & Simple numericals), Kirchoff’s laws (Statement & Explanation with diagram),Application of Kirchoff’s laws to Wheatstone bridge - Balanced condition of Wheatstone’s Bridge – Condition of Balance (Equation).
	2 <sup>nd</sup>	28.Electromagnetism – Definition & Concept,Force acting on a current carrying conductor placed in a uniform,magnetic field, Fleming’s Left Hand Rule
15 <sup>th</sup>	1 <sup>st</sup>	29. Faraday’s Laws of Electromagnetic Induction (Statement only),Lenz’s Law (Statement),Fleming’s Right Hand Rule,Comparison between Fleming’s Right Hand Rule and Fleming’s Left Hand Rule.
	2 <sup>nd</sup>	30. LASER & laser beam (Concept and Definition),Principle of LASER (Population Inversion & Optical Pumping),Properties & Applications of LASER,Wireless Transmission – Ground Waves, Sky Waves, Space Waves( Concept & Definition)

# LESSON PLAN

DEPARTMENT OF MATH AND SCIENCE , ITT, CHOUDWAR

SUBJECT: ENGG. PHYSICS PRACTICAL

Periods: 4 per week

SEMESTER: 1<sup>st</sup> & 2<sup>nd</sup>

NAME OF FACULTY: Pranabesh Bhanja

ACADEMIC YEAR.2020-2021

Semester From date:

To Date:

No. of weeks: 15

Week	Class Day	Practical Topics
1 <sup>st</sup>	1 <sup>st</sup>	1. To find the cross sectional area of a wire using a screw gauge.
	2 <sup>nd</sup>	2. To find volume of a solid cylinder using a Vernier Calipers(2)
2 <sup>nd</sup>	1 <sup>st</sup>	3. To find the thickness and volume of a glass piece using a screw gauge.
	2 <sup>nd</sup>	4. To find the thickness and volume of a glass piece using a screw gauge(2)
3 <sup>rd</sup>	1 <sup>st</sup>	5. To find volume of a solid cylinder using a Vernier Calipers.
	2 <sup>nd</sup>	6. To find volume of a solid cylinder using a Vernier Calipers.
4 <sup>th</sup>	1 <sup>st</sup>	7. To find volume of a solid cylinder using a Vernier Calipers.
	2 <sup>nd</sup>	8. To find volume of a solid cylinder using a Vernier Calipers.
5 <sup>th</sup>	1 <sup>st</sup>	9. To determine the radius of curvature of convex surface using a Spherometer(1).
	2 <sup>nd</sup>	10. To determine the radius of curvature of convex surface using a Spherometer(2).
6 <sup>th</sup>	1 <sup>st</sup>	11. To determine the radius of curvature of convex surface using a Spherometer(3).
	2 <sup>nd</sup>	12. To determine the radius of curvature of concave surface using a Spherometer(1).
7 <sup>th</sup>	1 <sup>st</sup>	13. To determine the radius of curvature of concave surface using a Spherometer(2).
	2 <sup>nd</sup>	14. To determine the radius of curvature of concave surface using a Spherometer(3).
8 <sup>th</sup>	1 <sup>st</sup>	15. To find the time period of a simple pendulum and determine acceleration due to gravity(1).

	2 <sup>nd</sup>	16. To find the time period of a simple pendulum and determine acceleration due to gravity(2).
9 <sup>th</sup>	1 <sup>st</sup>	17. To determine the angle of Prism(1).
	2 <sup>nd</sup>	18. To determine the angle of Prism(2).
10 <sup>th</sup>	1 <sup>st</sup>	19. To determine the angle of Minimum Deviation by I ~ D curve method(1).
	2 <sup>nd</sup>	20. To determine the angle of Minimum Deviation by I ~ D curve method(2).
11 <sup>th</sup>	1 <sup>st</sup>	21. To determine the angle of Minimum Deviation by I ~ D curve method(3).
	2 <sup>nd</sup>	22. To trace lines of force due to a bar magnet with North pole pointing North and locate the neutral points(1)
12 <sup>th</sup>	1 <sup>st</sup>	23. To trace lines of force due to a bar magnet with North pole pointing North and locate the neutral points(2)
	2 <sup>nd</sup>	24. To trace lines of force due to a bar magnet with North pole pointing North and locate the neutral points(3)
13 <sup>th</sup>	1 <sup>st</sup>	25. To trace lines of force due to a bar magnet with North pole pointing South and locate the neutral points(1).
	2 <sup>nd</sup>	26. To trace lines of force due to a bar magnet with North pole pointing South and locate the neutral points(2).
14 <sup>th</sup>	1 <sup>st</sup>	27. To trace lines of force due to a bar magnet with North pole pointing South and locate the neutral points(3).
	2 <sup>nd</sup>	28. To verify Ohm's Law by Ammeter – Voltmeter method(1).
15 <sup>th</sup>	1 <sup>st</sup>	29. To verify Ohm's Law by Ammeter – Voltmeter method(2).
	2 <sup>nd</sup>	30. To verify Ohm's Law by Ammeter – Voltmeter method(3).